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$\times 8.314 \times 10^{-5} \text{ (bar} \cdot \text{m}^3 \text{ mol}^{-1} \text{K}^{-1}) T$   
 $P = \frac{V}{V_1} \times 10^{-3} - (10 - x) 10^6$   
volume taken up by liquid Equating  $P$   
and  $P_{\text{vap}}$  we have  $x 18 \times 8.314 \times$   
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approximate solution can be found if  
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2. 3kT V m. Assume the temperature  
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