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514. BIG Idea Chemical reactions usually absorb or release energy. 15.1 Energy. MAIN Idea Energy can change form and flow, but it is always conserved. 15.2 Heat. MAIN Idea The enthalpy change for a reaction is the enthalpy of the products minus the enthalpy of the reactants. Chapter 15: Energy and Chemical Change

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Calorimeter: An insulated device used to measure the absorbed or Release of heat in chemical or physical

processes. b. Heat of Reaction: The enthalpy change For the chemical

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equation is exactly as it is written. c.

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applied on an object over a \_\_\_\_\_.  
force, distance. For work to be done,  
the following must happen. movement  
and movement in the same direction  
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Chapter 15: Energy and Chemical Change questionThermodynamics answerthe study of energy changes that occur during chemical reactions and changes in state questionChemical Potential Energy

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5. true 6. true 7. convection 8.  
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absorbed 12. more 13. a silver spoon;  
Silver is a better conductor of heat  
than wood. 14.

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